

Chemistry Of Dyes And Principle Of Dyeing By V A Shenai

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Dyes are normally water-soluble or water dispersible organic compounds that are capable of being absorbed into the substrate destroying the crystal structure of the substance. The dye molecules are usually chemically bonded to the surface and become a part of the material on which it is applied.

Chapter 8-DYES-THE CHEMISTRY AND APPLICATIONS GENERAL INTRODUCTION TO THE CHEMISTRY OF DYES 1. Principles of Colour Chemistry. 1.1 Basis for colour. Unlike most organic compounds, dyes possess colour because they 1) absorb light in the visible spectrum (400–700 nm), 2) have at least one chromophore (colour-bearing group), 3) have a conjugated system, i.e. a structure with alternating double and single bonds, and 4) exhibit resonance of electrons, which is a stabilizing force in organic compounds (Abrahart, 1977).

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V. A. Shenai, "Chemistry of Dyes and Principles of Dyeing ... 4.7.3. Principles of sulphur dye application. The principles of application of sulphur to cotton are the same as those for vat dyes. The process consists of four steps including reduction of the dye to make it soluble, application of the dye, i.e. dyeing, oxidation of the absorbed dye in the fibre and finally rinsing or quick washing. Sodium sulphide is a commonly used reducing agent.

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Chemistry Of Dyes And Principle Of Dyeing By V A Shenai Chemistry Of Dyes And Principle GENERAL INTRODUCTION TO THE CHEMISTRY OF DYES 1 ... Dyes developed for polyesters are known as disperse dyes In this case, the mechanism of coloration involves "dissolving" the dye in the polymer matrix to form a solid-solid

[DOC] Chemistry Of Dyes And Principle Of Dyeing By V A Shenai Chemistry and Industry Therefore, these dyes are mainly used to dye cotton fibres. Indigo is an important example of this type (Table 34.1). (viii) Mordant Dyes These dyes require an additional substance (generally a metal ion) for fixing to the fibre. These are used mainly for dyeing wool. The method involves the precipitation of certain

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Chemistry Of Dyes And Principle Of Dyeing By V A Shenai The unmasked dyes are highly colored and fluorescent ($\epsilon = 3\text{--}8 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$ and fluorescence quantum yields (ϕ) =40–85 % in the unbound state and in methanol). By stepwise and orthogonal protection of carboxylic and sulfonic acid groups a highly water-soluble caged red-emitting dye with two sulfonic acid residues was prepared.

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